Acta Cryst. (1964). 17, 1611

Crystal data (II) for some estrone-related compounds.* By Jean M. Ohrt, Barbara A. Haner, and Dorita A. Norton, Department of Biophysics, Roswell Park Memorial Institute, Buffalo, New York, U.S.A.

(Received 1 May 1964)

The lattice constants of eleven estrone-related compounds (Table 1) were determined on a General Electric XRD 5 X-ray diffraction unit equipped with a goniostat using $Cu\ K\alpha$ radiation. Space groups were established on the basis of systematic absences and optical activity.

All crystals were grown from solution. Flotation density measurements were used to calculate the number of molecules per unit cell. The high discrepancy between the measured (D_m) and calculated (D_x) densities of compound (3) is due to the crystallization of one molecule of ethanol per molecule of steroid. The density calculated taking the solvent of crystallization molecules into

consideration is 1.151 g.cm⁻³ which agrees with the measured density. All of the other measured and calculated densities also agree within the experimental error (4.0%).

Compounds (1) and (2) each crystallized in two forms. Both of the (a) forms had similar but non-isomorphous unit cells. Compounds (2) and (3) are isomers but do not form isomorphous crystals.

* This investigation was supported in part by P.H.S. grant CY-06183 from the National Cancer Institute, Public Health Service.

Table 1. Crystal data (II) for some estrone-related compounds

(1)(a), (b) 1,3,5(10)-Estratrien-3-ol-17-one

(2)(a), (b) 1,3,5(10)-Estratrien-3,17 α -diol

(3) 1,3,5(10)-Estratrien-3,17 β -diol

(4) 1,3,5(10)-Estratrien-3,17 β -diol 3-methyl ether

(5) 1,3,5(10)-Estratrien-3,17 β -diol 17-acetate

(6) 1,3,5(10)-Estratrien-17 α -ethinyl-3,17 β -diol 3-methyl ether

(7) 1.3.5(10)-Estratrien-3.17 α -diol diacetate

(8) 1,3,5(10)-Estratrien-3-ol-17-one trimethyl acetate

(9) 1,3,5(10)-Estratrien-3,17 α -diol 3-trimethyl acetate

(10) 1,3,5(10)-Estratrien-3,17 β -diol dipropionate

(11) 1,3,5(10)-Estratrien-3,17 β -diol 3-benzoate

	l(a)	1(b)	2(a)	2(b)	3	4	5
Formula	$\mathrm{C_{18}H_{22}O_2}$	$\mathrm{C_{18}H_{22}O_{2}}$	$\mathrm{C_{18}H_{24}O_{2}}$	$\mathrm{C_{18}H_{24}O_{2}}$	$\mathrm{C_{18}H_{24}O_{2}}$	$C_{19}H_{26}O_{2}$	$\mathrm{C_{20}H_{26}O_3}$
Mol. wt.	$270\overline{.37}$	$270\overline{.37}$	$272\overline{.39}$	$272 \cdot 39$	$272\overline{\cdot}39$ "	286.42	314.43
D_m (g.cm ⁻³)	1.17_{6}	1.24_{8}	1.20	1.20_{7}	$1 \cdot 14_{8}$	1.20_{6}	1.20_{2}
D_x (g.cm ⁻³)	1.220	1.249	1.190	1.190	0.985	1.186	$1\cdot22\overline{0}$
Space group	$P2_1$	$P2_{1}2_{1}2_{1}$	$P2_1$	C2	$P2_{1}2_{1}2$	$P2_{1}2_{1}2$	$P22_{1}2$
Z (calc.)	4	4	4	4	4	8 -	4
a (Å)†	9.297	10.051	9-127	19.212	$12 \cdot 241$	14.004	13.803
b (Å)†	$23 \cdot 317$	18.440	23.293	$7 \cdot 132$	$23 \cdot 215$	35.458	16.912
c (Å) \dagger	$7 \cdot 662$	7.755	7.238	13.422	$6 \cdot 466$	6.470	7.331
β (°)	$112 \cdot 26$		98.78	$124 \cdot 25$			
$Volume (Å^3)$	1471	1437	1521	1520	1837	3213	1711
Solvent	Methanol	$\mathbf{Acetone}$	Methanol	Methanol	95% Ethanol	Heptane	95% Ethanol

 $+ \pm 0.007.$

Table 1 (cont.)

	6	7	8	9	10	11
Formula	$\mathrm{C_{21}H_{28}O_2}$	$\mathrm{C_{22}H_{28}O_4}$	$\mathrm{C_{23}H_{30}O_3}$	$\mathrm{C_{23}H_{32}O_3}$	$\mathrm{C}_{24}\mathrm{H}_{32}\mathrm{O}_4$	$\mathrm{C_{25}H_{28}O_3}$
Mol. wt.	310.44	$\overline{356.47}$	$354 \cdot 49$	$356\overline{51}$	384.52	376.50
D_m (g.cm ⁻³)	1.22_{2}	1.20_{3}	1.16_{8}	$1 \cdot 14_{4}$	1.20_{9}	$1 \cdot 21_4$
D_x (g.cm ⁻³)	1.217	1.246	1.175	1.153	1.182	1.251
Space group	$P2_1$	$P2_{1}2_{1}2_{1}$	$P2_1$	$P2_1$	$P2_1$	$P2_{1}2_{1}2_{1}$
Z (calc.)	4	4	2^{-}	2	4	4
a (Å)†	6.998	8.663	20.704	16.805	23.038	13.331
b (Å)†	39.737	$29 \cdot 390$	7.318	9.714	9.057	23.846
$c(A)\dagger$	6.871	7.460	6.781	6.356	10.352	6.286
β (°)	117.58		102.51	98.36	95.36	
$\dot{\text{Volume}}$ (Å ³)	1694	1899	1003	1027	2160	1998
Solvent	Heptane	Methanol	Propanol	Heptane- acetone	Acetone- methanol	95% Ethanol

 $† \pm 0.007.$